

Solution to written exam in QUANTUM PHYSICS F0047T

Examination date: 2010-01-15

1. (a) i. $\hat{\Pi}C\left(\sin\left(\frac{\pi x}{L}\right) + \sin\left(\frac{3\pi x}{L}\right)\right) = C\left(\sin\left(\frac{-\pi x}{L}\right) + \sin\left(\frac{-3\pi x}{L}\right)\right) = -C\left(\sin\left(\frac{\pi x}{L}\right) + \sin\left(\frac{3\pi x}{L}\right)\right)$, the eigenvalue is -1
 - ii. $\hat{\Pi}Ce^{-a\sqrt{x^2+y^2+z^2}} = CCe^{-a\sqrt{(-x)^2+(-y)^2+(-z)^2}} = Ce^{-a\sqrt{x^2+y^2+z^2}}$, the eigenvalue is +1
 - iii. $\hat{\Pi}Cf(r)(\cos(\theta) + \cos^3(\theta))e^{i\phi} = Cf(r)(\cos(\pi - \theta) + \cos^3(\pi - \theta))e^{i(\phi+\pi)} = Cf(r)(-\cos(\theta) - \cos^3(\theta))(-e^{i\phi}) = Cf(r)(\cos(\theta) + \cos^3(\theta))e^{i\phi}$, the eigenvalue is +1
 - (b) i. $\hat{\Pi}(2\psi_+(x, y, z) + 3\psi_-(x, y, z)) = +2\psi_+(x, y, z) - 3\psi_-(x, y, z) \neq \lambda(2\psi_+(x, y, z) + 3\psi_-(x, y, z))$, not an eigenfunction.
 - ii. $\hat{\Pi}^2(2\psi_+(x, y, z) + 3\psi_-(x, y, z)) = \hat{\Pi}(+2\psi_+(x, y, z) - 3\psi_-(x, y, z)) = 2\psi_+(x, y, z) + 3\psi_-(x, y, z)$, an eigenfunction with eigenvalue +1.
 - iii. $\hat{\Pi}e^{-ikx} = e^{+ikx} \neq e^{-ikx}$ not an eigenfunction and neither is e^{ikx} . We can however form linear combinations that have parity. The function $e^{ikx} - e^{-ikx}$ has parity $\hat{\Pi}(e^{+ikx} - e^{-ikx}) = e^{-ikx} - e^{+ikx} = -1(e^{+ikx} - e^{-ikx})$ with eigenvalue -1. The function $e^{ikx} + e^{-ikx}$ has parity $\hat{\Pi}(e^{+ikx} + e^{-ikx}) = e^{-ikx} + e^{+ikx} = +1(e^{+ikx} + e^{-ikx})$ with eigenvalue +1.
2. The eigenfunctions and eigenvalues of the free-particle Hamiltonian are found by solving the time-independent Schrödinger equation

$$-\frac{\hbar^2}{2m} \frac{d^2 u(x)}{dx^2} + V(x)u(x) = Eu(x),$$

with $V(x)$ zero everywhere. Thus, the eigenvalue equation reads

$$\frac{d^2 u(x)}{dx^2} + k^2 u(x) = 0,$$

where $k^2 = 2mE/\hbar^2$. The eigenfunctions are given by the plane waves e^{ikx} and e^{-ikx} , or linear combinations of these, as *e.g.* $\sin kx$ and $\cos kx$.

- (a) The wave function of the particle at $t = 0$ is given by

$$\psi(x, 0) = \cos^3(kx) + \sin^5(kx).$$

This is not an eigenfunction in itself but it can be written as using the Euler relations

$$\psi(x, 0) = \left(\frac{e^{ikx} + e^{-ikx}}{2}\right)^3 + \psi(x, 0) = \left(\frac{e^{ikx} - e^{-ikx}}{2i}\right)^5 = \quad (1)$$

$$\frac{1}{8} \left(e^{i3kx} + 3e^{ikx} + 3e^{-ikx} + e^{-i3kx}\right) + \quad (2)$$

$$\frac{1}{32i} \left(e^{i5kx} - 5e^{i3kx} + 10e^{ikx} - 10e^{-ikx} + 5e^{-i3kx} - e^{-i5kx}\right) = \quad (3)$$

$$\frac{3}{4} \cos(kx) + \frac{1}{4} \cos(3kx) + \frac{1}{16} \sin(5kx) - \frac{5}{16} \sin(3kx) + \frac{10}{16} \sin(kx) \quad (4)$$

Thus, $\psi(x, 0)$ can be written as a superposition of plane waves with three different values of $k_1 = k$, $k_2 = 3k$ and $k_3 = 5k$

- (b) The energy of a plane wave e^{ikx} is given by $E = \hbar^2 k^2 / 2m$. Thus, the energy of $e^{ik_1 x}$ (or $e^{-ik_1 x}$) is $E_1 = \hbar^2 k^2 / 2m$ and the energy of $e^{ik_2 x}$ (or $e^{-ik_2 x}$) is $E_2 = \hbar^2 k_2^2 / 2m = 9\hbar^2 k^2 / 2m$. and the energy of $e^{ik_3 x}$ (or $e^{-ik_3 x}$) is $E_3 = \hbar^2 k_3^2 / 2m = 25\hbar^2 k^2 / 2m$.
- (c) The function $u(x) = e^{ikx}$ is a solution to the the time-independent Schrödinger equation. The corresponding solutions to the time-dependent Schrödinger equation are given by $u(x)T(t)$, with $T(t) = e^{-iEt/\hbar}$. Therefore, $u(x)T(t) = e^{i(kx - Et/\hbar)}$. A sum of solutions of this form is also a solution, since the Schrödinger equation is linear. This means that if $\psi(x, 0)$ is given by equation (4), then the time dependent solution is given by

$$\psi(x, t) = \frac{1}{8} (e^{i3kx} + e^{-i3kx}) e^{-iE_2 t/\hbar} + \frac{3}{8} (e^{ikx} + e^{-ikx}) e^{-iE_1 t/\hbar} + \quad (5)$$

$$\frac{1}{32i} (e^{i5kx} - e^{-i5kx}) e^{-iE_3 t/\hbar} - \frac{5}{32i} (e^{i3kx} - e^{-i3kx}) e^{-iE_2 t/\hbar} + \frac{10}{32i} (e^{ikx} - e^{-ikx}) e^{-iE_1 t/\hbar} = \quad (6)$$

$$(7)$$

where

$$E_1 = \frac{\hbar^2 k^2}{2m} \quad \text{and} \quad E_2 = \frac{9\hbar^2 k^2}{2m} \quad \text{and} \quad E_3 = \frac{25\hbar^2 k^2}{2m} \quad (8)$$

3. (a) The parity of a hydrogen eigenfunction $\psi_{nlm_l}(\mathbf{r})$ is given by $(-1)^l$. The given wave function $\Psi(\mathbf{r})$ is a mixture of eigenfunctions of different parity. Hence $\Psi(\mathbf{r})$ cannot have a definite parity.
- (b) The probability is given by the absolute square of the coefficients. The probabilities are (in order) $\frac{4}{15}, \frac{9}{15}, \frac{1}{15}, \frac{1}{15}$. as a check they sum up to 1 as they should do.
- (c) The energy of a single state is given by: $E_n = -\frac{13.56}{n^2}$ eV. The expectation value is given by $\langle E \rangle = \frac{4}{15}(-\frac{13.56}{1^2}) + \frac{9}{15}(-\frac{13.56}{2^2}) + \frac{1}{15}(-\frac{13.56}{3^2}) + \frac{1}{15}(-\frac{13.56}{3^2}) = -13.56(\frac{4}{15} + \frac{9}{60} + \frac{1}{135} + \frac{1}{135}) = -5.851$ eV

The operator \mathbf{L}^2 has eigenvalues $\hbar^2 l(l+1)$. The expectation value is given by

$$\langle \mathbf{L}^2 \rangle = \frac{4}{15} \cdot 0 + \frac{9}{15} \cdot 0 + \frac{1}{15} (\hbar^2 1(1+1)) + \frac{1}{15} (\hbar^2 2(2+1)) = \frac{8}{15} \hbar^2$$

The operator L_z has eigenvalues $\hbar m_l$. The expectation value is given by

$$\langle L_z \rangle = \frac{4}{15} \cdot 0 + \frac{9}{15} \cdot 0 + \frac{1}{15} \cdot 0 + \frac{1}{15} (\hbar 2) = \frac{2}{15} \hbar$$

4. The eigenfunctions of the infinite square well are (Physics handbook)

$$\psi_n(x) = \sqrt{\frac{2}{a}} \sin \frac{n\pi x}{a} \quad \text{and the eigenenergies are } E_n = \frac{n^2 \pi^2 \hbar^2}{2ma^2} \quad \text{where } n = 1, 2, 3, \dots$$

The correction to the eigenenergies due to perturbation is given by:

$$E_n^1 = \langle H_1 \rangle \quad \text{where } H_1 \text{ is the deviation in the potential from the infinite square well.}$$

$$E_n^1 = \int_0^{a/2} \frac{2\epsilon}{a} \sin^2 \frac{n\pi x}{a} dx = \int_0^{a/2} \frac{2\epsilon}{a^2} \left(1 - \cos \frac{2n\pi x}{a}\right) dx = \frac{\epsilon}{a} \left[x - \frac{a}{2n\pi} \sin \frac{2n\pi x}{a} \right]_0^{a/2} = \frac{\epsilon}{2}$$

This is the same for all n . The corrections for $n=1$ and $n=2$ are of interest (**answer to a**).

$$E_1^1 = E_2^1 = \frac{\epsilon}{2} = 0.055 \text{ eV.}$$

The two lowest eigenenergies are

$$E_n = \frac{n^2 \pi^2 \hbar^2}{2ma^2} = \frac{n^2 h^2}{8ma^2} \quad [n = 1] \quad E_1 = 6.02465 \cdot 10^{-20} \text{ J} = 0.376028 \text{ eV} \quad \text{and} \quad E_2 = 1.504114 \text{ eV}$$

The new energies for the two lowest levels are:

$$E_1^* = 0.376028 + 0.055 = 0.431028\text{eV} \quad \text{and} \quad E_2^* = 1.504114 + 0.055 = 1.559114\text{eV}$$

The case of $\epsilon = 1.08$ eV. This energy is much higher than the groundstate energy of the unperturbed model. You cannot use perturbation here. The model is more like a square well of width $a/2$. The answer of such a perturbation calculation would not give a trustworthy result.

5. The rotational energy of a molecule is given by

$$E_l = \frac{\hbar^2}{2I}l(l+1)$$

The emitted photon energy for a transition $l+1$ to l is given by

$$E_{\text{photon}} = \frac{\hbar^2}{2I}((l+1)(l+2) - l(l+1)) = \frac{\hbar^2}{2I}(l^2 + 3l - 2 - l^2 - l) = \frac{\hbar^2(l+1)}{I}$$

This is the energy for one line and there will be a set of lines all for different l . The separation between two adjacent lines in energy will be

$$E_{\text{photon},l+2} - E_{\text{photon},l+1} = \frac{\hbar^2(l+2)}{I} - \frac{\hbar^2(l+1)}{I} = \frac{\hbar^2}{I}$$

The energy of a photon is $E_{\text{photon}} = \frac{hc}{\lambda}$ and hence

$$\frac{\hbar^2}{I} = hc \left(\frac{1}{\lambda_1} - \frac{1}{\lambda_2} \right) = 1.2398 \cdot 10^{-6} \left(\Delta \frac{1}{\lambda} \right) \text{eV m} = 1.2398 \cdot 10^{-6} \cdot 20.68 \text{cm}^{-1} = 2.564 \cdot 10^{-3} \text{eV}$$

$$I = \frac{(1.055 \cdot 10^{-34})^2}{2.564 \cdot 10^{-3} \cdot 1.602 \cdot 10^{-19}} = 2.71 \cdot 10^{-47} \text{kg m}^2$$

The moment of inertia is $I = mr^2$ where $m = m_H m_{Cl} / (m_H + m_{Cl}) = 35/36 m_H$ and r is the average separation. $r = \sqrt{\frac{I}{m}} = \sqrt{\frac{36 \cdot 2.71 \cdot 10^{-47}}{35 \cdot 1.673 \cdot 10^{-27}}} = 1.29 \cdot 10^{-10} \text{m}$

$$E_l = \frac{2.56 \cdot 10^{-3}}{2} l(l+1) \text{eV}$$

$$E_0 = 0 \text{eV},$$

$$E_1 = 2.56 \cdot 10^{-3} \text{eV} = 4.10 \cdot 10^{-22} \text{J},$$

$$E_2 = 7.68 \cdot 10^{-3} \text{eV} = 1.23 \cdot 10^{-21} \text{J},$$

$$E_3 = 15.4 \cdot 10^{-3} \text{eV} = 2.47 \cdot 10^{-21} \text{J},$$

$$E_3 = 25.6 \cdot 10^{-3} \text{eV} = 4.10 \cdot 10^{-21} \text{J}.$$