# LULEÅ UNIVERSITY OF TECHNOLOGY Applied Physics

| Course code      | F7035T        |
|------------------|---------------|
| Examination date | 2019-05-29    |
| Time             | 09.00 - 14.00 |

| Examination in:  | STATISTICAL | Physics | AND | Thermodynamics           |
|------------------|-------------|---------|-----|--------------------------|
| Teacher on duty: | Hans Weber  |         |     | Tel: (49)2088, Room E163 |
| Examiner: Hans   | Weber       |         |     | Tel: (49)2088, Room E163 |

Allowed aids: Fysikalia/Fysika, Physics Handbook, Beta, calculator, COLLECTION OF FORMULAE

Define notations and motivate assumptions and approximations. Present the solutions so that they are easy to follow.

Total number of problems: 5. Maximum number of point is 15 p. 7.5 points is required to pass the examination. Grades 3: 7.5, 4: 10.0, 5: 12.0

## 1. Helium ${}^{3}He$

Helium  ${}^{3}He$  has spin =  $\frac{1}{2}$  and may at low temperatures to a good approximation be described as an ideal Fermi gas. At these low temperatures  ${}^{3}He$  is in the liquid phase with a density of  $\rho = 83$  kg m<sup>-3</sup>.

- a) Determine the Fermi temperature  $T_F$  and also the specific heat  $C_v$  of  ${}^{3}He$  at T=0.2 K.
- b) Can you still use the approximations you did in a) if the temperature where say 2-3 K? If not why? If yes why?

(3p)

### 2. Adsorption

Consider an ideal gas (temperature T, chemical potential  $\mu$ ) in contact with a surface with N adsorption sites. Each adsorption site may be occupied by 0, 1 or 2 gas molecules. The energy of a vacant site is zero, the energy with one adsorbed molecule is  $-\epsilon$  and the energy with two adsorbed molecules is  $-\frac{3}{2}\epsilon$ .  $\epsilon$  can be positive or negative. There is no interaction between molecules at different adsorption sites.

- a) Calculate the grand canonical partition function for a fixed number N of adsorption sites.
- b) Use the grand canonical partition function to derive the mean number of adsorbed particles per site  $\langle n \rangle$  and the mean internal energy per site  $\langle u \rangle$  as a function of temperature T,  $\mu$  and  $\epsilon$ .

(3p)

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#### 3. The three dimensional Ising–model in the mean field approximation

The three dimensional Ising model on a cubic lattice has the following 'Hamiltonian'

$$H = -J \sum_{\langle i,j \rangle} s_i s_j,$$

where the classical spins s have the following states +1 and -1. The spins  $s_i$  interact with their nearest neighbours. Let J = 1 and the system will have a ferro magnetic ground state, ie the magnetisation at temperature  $\tau = 0$  is  $\langle m \rangle = \frac{1}{L^3} \sum_i s_i = 1$ .

As the temperature is raised the magnetisation disappears at a specific temperature the Curie temperature  $\tau_c$ . As the temperature approaches  $\tau_c$  from below the magnetization goes to zero according to  $m \propto (\tau_c - \tau)^{\beta}$ .

Within the mean field approximation calculate the exponent  $\beta$  for the magentisation.

$$(\tanh(x) \approx x - x^3/3 \text{ for small } x).$$
 (3p)

#### 4. Harmonic oscillator

A three dimensional harmonic oscillator has energy levels

$$\epsilon_{n_1,n_2,n_3} = (n_1 + n_2 + n_3 + \frac{3}{2}) \hbar \omega$$

where  $n_1, n_2, n_3$  är are integers from 0 to  $\infty$ .

- a) At what temperature is the probability for the oscillator to be in a state of energy  $\frac{3}{2}\hbar\omega$  or  $\frac{5}{2}\hbar\omega$  the same?
- **b)** How large is this probability ?

(3p)

#### 5. Diatomic molecule

An ideal gas consists of N identical molecules. Each molecule consists of two atoms with the following rotational energy levels:  $E(j) = j(j+1)\frac{\hbar^2}{2I}$ , j = 0, 1, 2, ... Where I is the moment of inertia. Each level is (2j+1) times degenerate. For low temperatures determine to lowest order in temperature the contribution to  $C_v$  from the rotational degrees of freedom.

(3p)

# GOOD LUCK !